

Photoaddition Reaction of 1,2-Dialkyl-indoles and -pyrroles to 1-Methyl-2-pyridone *via* Proton Transfer from the 2-Methylene Group of the Indole or Pyrrole

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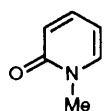
On irradiation, 1,2-dialkyl-indoles and -pyrroles undergo 1,2- and 1,4-addition to 1-methyl-2-pyridone in a reaction involving single electron transfer followed by proton transfer from the 2-alkyl group of the indole or pyrrole; bonding takes place between C-3 of the indole, or C-1 of the 2-alkyl group of the indole or pyrrole and C-4 or C-6 of the pyridone, respectively.

Recently we have reported that irradiation of 1-methyl-2-pyridone **1** in the presence of indoles and pyrroles primarily gives rise to the 1,2- and 1,4-addition of the annular 2-H or 3-H bond of the π -excessive heterocycles to the 2-pyridone nucleus yielding 4-arylated 3,4-dihydropyridin-2(1H)-ones and 6-arylated 3,6-dihydropyridin-2(1H)-ones.¹ The photoreaction is presumed to proceed *via* a single electron transfer (SET) process, followed by proton transfer and then combination of the resulting radicals. The presence of a pyrrolyl or an indolyl

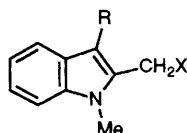
NH group is essential to the photoaddition reaction because it acts as a proton source in the proton transfer process, which is similar to the photoaddition reaction of pyrrole to benzene² and naphthalene.³ We report here the photoaddition reaction of 1,2-dialkyl-indole **2-5** and -pyrrole derivatives **6, 7** to 1-methyl-2-pyridone **1**, in which the methylene (or methyl) group at the 2-position of the indole and the pyrrole ring acts as a proton source instead of the NH group.

The photoreactions were performed on degassed dioxane or acetonitrile solutions of the π -excessive heterocycles (1.0 mol dm⁻³) and **1** (0.05 mol dm⁻³) by external irradiation with a 400 W high-pressure mercury lamp through a Pyrex filter for 15 h at ambient temperature. Table 1 records product distribution and yields.

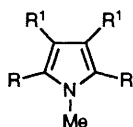
Irradiation of **1** with 1,2-dimethylindole **2** gave three 1:1 adducts **8, 9** and **10**[†] in 8.5, 10 and 4% yields, respectively, based on consumed **1**. The former two, **8** and **9**, were attributable to the 1,2- and 1,4-addition of the nuclear 3-H bond of **2** to **1**, in analogy with the photoreaction between



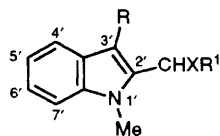
1



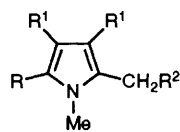
2; X = R = H
3; X = H, R = Me
4; X = CO₂Et, R = H
5; X = CN, R = H



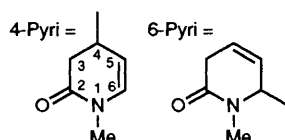
6; R = Me, R¹ = H
7; R = R¹ = Me



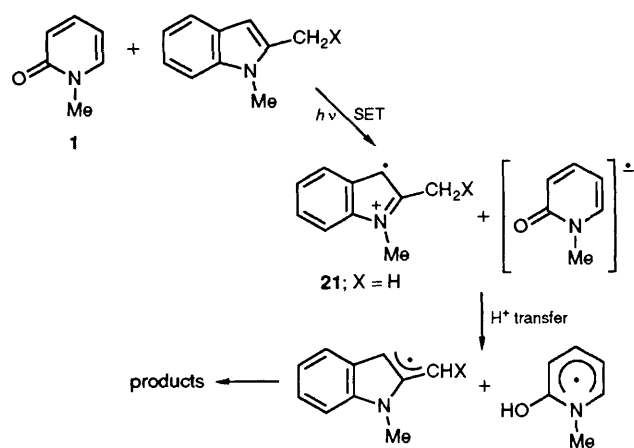
8; X = R¹ = H, R = 4-Pyri
9; X = R¹ = H, R = 6-Pyri
10; X = R = H, R¹ = 4-Pyri
11; X = H, R = Me, R¹ = 6-Pyri
12; X = CO₂Et, R = 6-Pyri, R¹ = H
13; X = CO₂Et, R = H, R¹ = 4-Pyri
14; X = CN, R = 6-Pyri, R¹ = H
15; X = CN, R = H, R¹ = 4-Pyri
16; X = CN, R = H, R¹ = 6-Pyri



17; R = Me, R¹ = H, R² = 4-Pyri
18; R = Me, R¹ = H, R² = 6-Pyri
19; R = R¹ = Me, R² = 4-Pyri
20; R = R¹ = Me, R² = 6-Pyri



[†] All new compounds were characterized by ¹H NMR, IR and mass spectra. *Selected spectroscopic data*: for **8**: ¹H NMR (CDCl₃) δ 2.36 (s, 3H, 2'-Me), 2.62 (ddd, 1H, *J* 16.5, 6.7 and 1.2 Hz, 3-H), 2.90 (dd, 1H, *J* 16.5 and 14.7 Hz, 3-H'), 3.15 (s, 3H, 1-Me), 3.66 (s, 3H, 1'-Me), 4.12 (dddd, 1H, *J* 14.7, 6.7, 3.7 and 3.1 Hz, 4-H), 5.32 (ddd, 1H, *J* 7.3, 3.7 and 1.2 Hz, 5-H), 6.16 (dd, 1H, *J* 7.3 and 3.1, 6-H), 7.04 (dd, 1H, *J* 7.9 and 7.9 Hz, 5'-H), 7.16 (dd, 1H, *J* 7.9 and 7.9 Hz, 6'-H), 7.27 (d, 1H, *J* 7.9 Hz, 7'-H), 7.60 (d, 1H, *J* 7.9 Hz, 4'-H); IR (CHCl₃): 1650 cm⁻¹ (C=O). For **10**: ¹H NMR (CDCl₃) δ 2.25 (s, 3H, 2'-Me), 2.67-2.88 (m, 2H, 3-H₂), 2.93 (dd, 1H, *J* 14.5 and 8.5 Hz, one of CH₂), 3.07 (s, 3H, 1-Me), 3.24 (dd, 1H, *J* 14.4 and 4.8 Hz, one of CH₂), 3.66 (s, 3H, 1'-Me), 4.07 (m, 1H, 6-H), 5.69 (ddd, 1H, *J* 7.3, 4.3 and 2.4 Hz, 5-H), 5.76 (ddd, 1H, *J* 10.4, 4.3 and 2.4 Hz, 4-H), 7.10 (dd, 1H, *J* 7.9 and 7.9 Hz, 5'-H), 7.20 (dd, 1H, *J* 7.9 and 7.9 Hz, 6'-H), 7.26 (d, 1H, *J* 7.9, 7'-H), 7.51 (d, 1H, *J* 7.9, 4'-H); IR (CHCl₃): 1620 cm⁻¹ (C=O).



indoles and **1**.¹ The last product **10** came from the 1,2-addition of the C–H bond of the 2-methyl group of **2** to **1**. This was a new type of addition product in the photoreaction between π -excessive heterocycles and aromatic compounds. Similarly, indoles **4** and **5** gave the products attributable to both annular (**12**, **14**) and lateral (**13**, **15**, **16**) addition. On the other hand, the introduction of a methyl group in the 3-position of the indole ring showed significant regiocontrol for the lateral addition. Indole **3** afforded only the side-chain addition product **11**. In addition, the photoreaction of pyrroles **6** and **7** also occurred exclusively by the lateral addition to give adducts **17** and **18**, and **19** and **20**, respectively.

Holze and Hamann⁴ have reported that the electron distribution of the 1,2-dimethylindole radical cation is as shown in the resonance structure **21**. This indicates the extreme probability for proton release from the 2-methyl

Table 1 Photoaddition products from irradiation of 1-methyl-2-pyridone **1** with indoles **2–5** and pyrroles **6** and **7**

Indole or Pyrrole	Solvent ^a	Products and yields (%) ^b		
		Annular	Lateral	
2	D	8 (8.5)	9 (10)	10 (4)
	A	8 (8.5)	9 (9.5)	10 (2)
3	D			11 (24)
4	D	12 (14)		13 (22.5)
5	A	14 (22)		15 (20) 16 (4)
6	D			17 (20) 18 (27)
7	D			19 (44) 20 (39)

^a D, dioxane; A, acetonitrile. ^b Isolated yields based on consumed **1**.

group on the side chain. Therefore, the photoaddition reaction was presumed to proceed by way of proton transfer from the 2-methyl (or methylene) group of the pyrrole or indole to the ketyl radical of **1**, following an SET process, as shown in Scheme 1. The present reaction provides the first example of the formation of the side-chain addition products in photoaddition reactions of π -excessive heterocycles to aromatic compounds.

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References

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